Two-Photon Direct Frequency Comb Spectroscopy of Potassium

Mike Rowan

Why do we study atoms?
 What is atomic spectroscopy?
 What is a frequency comb?
 Why Potassium?
 What have I done this summer?

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Atoms are relatively simple

 Good theoretical understanding of atoms
 We can make models and calculations

 We can control them well by use of lasers

 Extreme accuracy of measurements

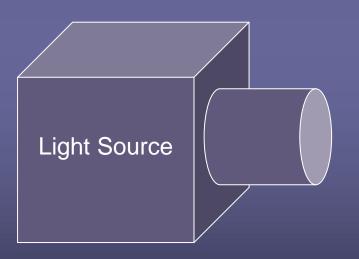
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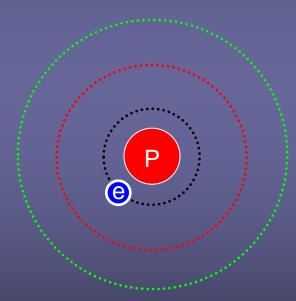
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 Extreme accuracy of measurements
 - High-precision experiments provide tests of our theories and let us learn more about fundamental physics like parity violation, fundamental constants, general relativity, the weak interaction, quantum electrodynamics, etc.

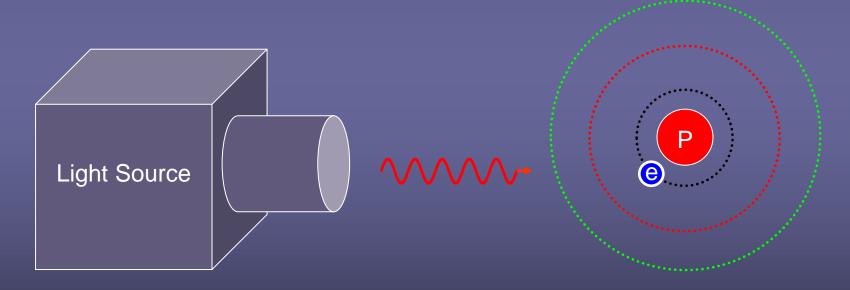
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- Spectroscopy the use of light emission and absorption to study matter
- First, we excite an atom using light
- Energy of a photon proportional to frequency (color); *E=hv*

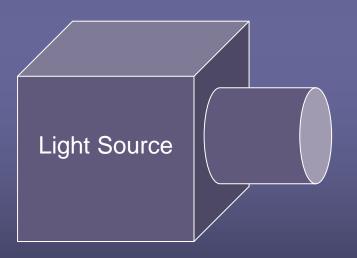


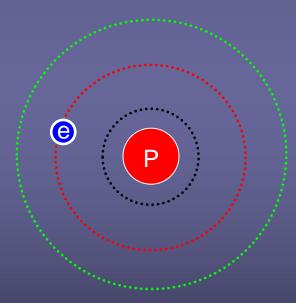


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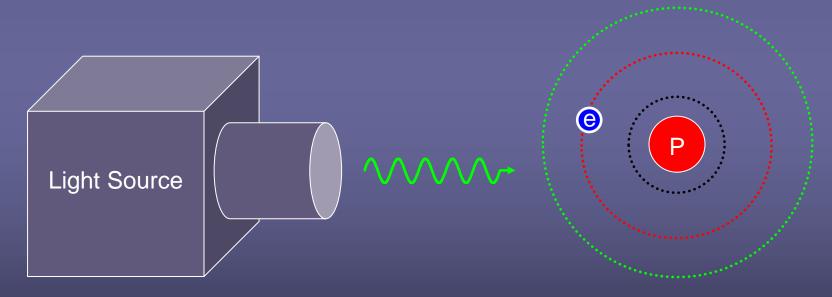


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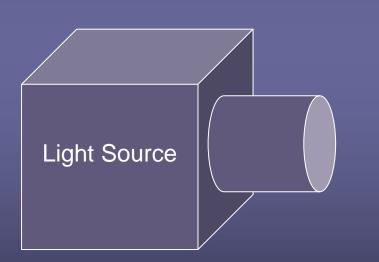


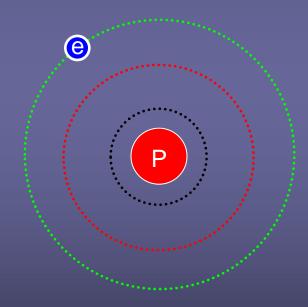


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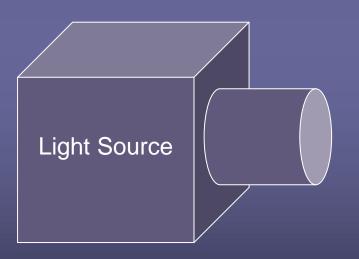
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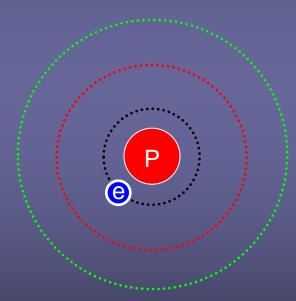




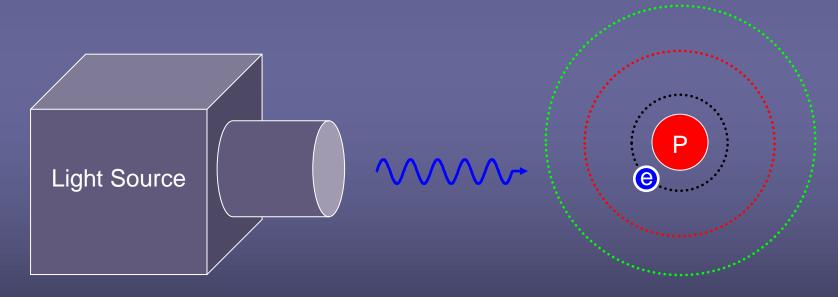
Two photon transition!

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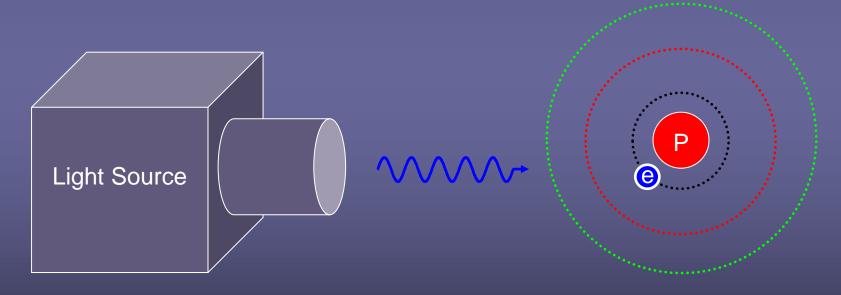




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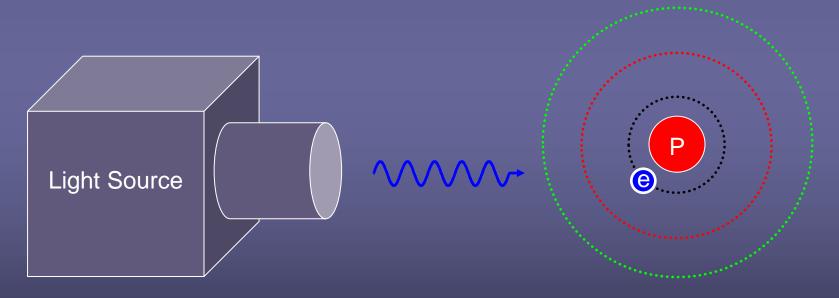


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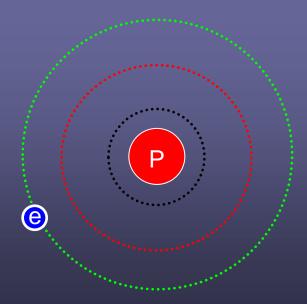
No Transition

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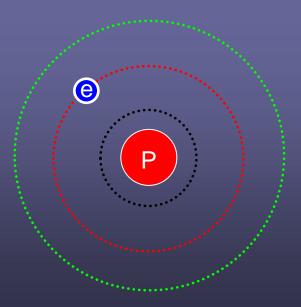
Light must have the right frequency to result in a transition

- An atom in the excited state then decays, producing fluorescence
- We detect the light using a photomultiplier tube (PMT)
- Intensity of the light indicates the population of atoms in the excited state





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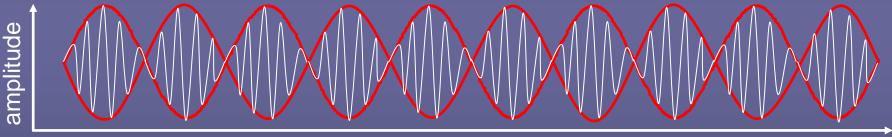
- We measure the frequency (v) of the light that causes transitions
- The strongest transitions are usually in the visible region of the optical spectrum
- However, frequencies of visible light are very high – on the order of

400,000,000,000,000 Hz

- Modern electronics cannot count frequencies this high
- However, we can use interference to produce a measurable frequency

Interference

- The rate of a beat is the difference of two frequencies it results from the interference of two slightly different frequencies
- We hear this as a periodic variation in volume



time

- If you know one frequency and you know the beat frequency, then you can determine the second frequency
- By interfering frequencies of visible light, we get a beat frequency that is in the radio range this is measurable

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How do we measure frequencies of light?

• We make a "light ruler"

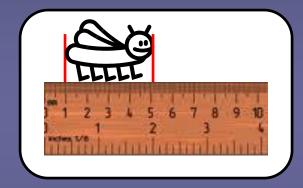


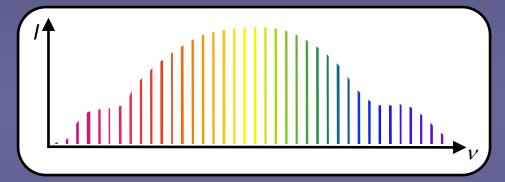
Ruler: Number of ticks Offset Spacing between ticks

- This structure is called an optical frequency comb a set of equidistantly spaced spectral lines
- By interfering different colors in the comb, we can produce radio frequencies that we can control

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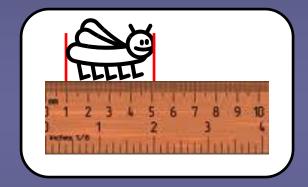


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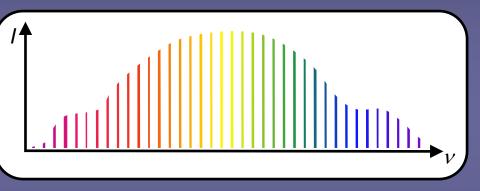
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Comb Equation: $v_n = n f_{rep} + f_0$

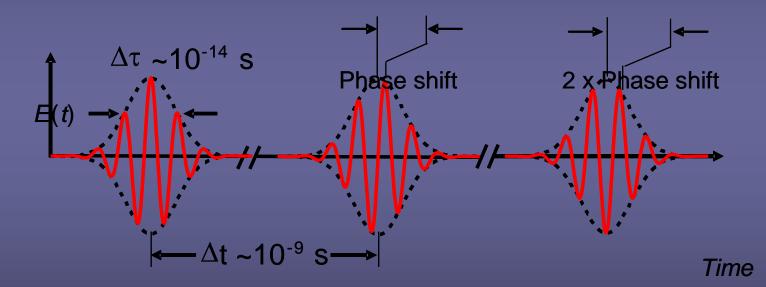


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How do we make a frequency comb?

• The comb is produced by a series of ultrashort pulses



- Phase coherence of the pulses leads to interference and the generation of an optical frequency comb
- Pulses are produced by a modelocked laser

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Why potassium?

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- Since it is simple, theory is good
- Potassium has not been studied as much as the other alkalis

The Periodic Table of the Elements																	
1 H Hydrogen 1.00794																	2 He Helium 4.003
3	4											5	6	7	8	9	10
Li	Be											В	C	Ν	0	F	Ne
Lithium 6.941	Beryllium 9.012182											Boron 10.811	Carbon 12.0107	Nitrogen 14.00674	Oxygen 15.9994	Fluorine 18.9984032	Neon 20.1797
11	12											13	14	15	16	17	18
Na	Mg											Al	Si	Р	S	Cl	Ar
Sodium 22.989770	Magnesium 24.3050											Aluminum 26.981538	Silicon 28.0855	Phosphorus 30.973761	Sulfur 32.066	Chlorine 35.4527	Argon 39.948
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Со	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Potassium 39.0983	Calcium 40.078	Scandium 44.955910	Titanium 47.867	Vanadium 50.9415	Chromium 51.9961	Manganese 54.938049	Iron 55.845	Cobalt 58.933200	Nickel 58.6934	Copper 63.546	Zinc 65.39	Gallium 69.723	Germanium 72.61	Arsenic 74.92160	Selenium 78.96	Bromine 79.904	Krypton 83.80
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те	Ι	Xe
Rubidium 85.4678	Strontium 87.62	Yttrium 88.90585	Zirconium 91.224	Niobium 92.90638	Molybdenum 95.94	Technetium (98)	Ruthenium 101.07	Rhodium 102.90550	Palladium 106.42	Silver 107.8682	Cadmium 112.411	Indium 114.818	Tin 118.710	Antimony 121.760	Tellurium 127.60	Iodine 126.90447	Xenon 131.29
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba	La	Hf	Та	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Ро	At	Rn
Cesium 132.90545	Barium 137.327	Lanthanum 138.9055	Hafnium 178.49	Tantalum 180.9479	Tungsten 183.84	Rhenium 186.207	Osmium 190.23	Iridium 192.217	Platinum 195.078	Gold 196.96655	Mercury 200.59	Thallium 204.3833	Lead 207.2	Bismuth 208.98038	Polonium (209)	Astatine (210)	Radon (222)
87	88	89	104	105	106	107	108	109	110	111	112	113	114				
Francium	Radium	Ac	Rf Rutherfordium	Db Dubnium	Sg	Bh Bohrium	Hassium	Mt Meitnerium									
(223)	(226)	(227)	(261)	(262)	(263)	(262)	(265)	(266)	(269)	(272)	(277)						
				58	59	60	61	62	63	64	65	66	67	68	69	70	71
				Cerium	Praseodymium	Nd Neodymium	Pm	Sm Samarium	Europium	Gadolinium	Tb Terbium	Dy Dysprosium	Holmium	Erbium	Tm Thulium	Yb Ytterbium	Lu
				140.116	140.90765	144.24	(145)	150.36	151.964	157.25	158.92534	162.50	164.93032	167.26	168.93421	173.04	174.967
				90	91	92 • •	93	94 D	95	96	97	98	99	100	101	102	103
				Th	Pa Protactinium	U	Np Neptunium	Pu	Am Americium	Curium	Bk Berkelium	Cf	Es	F'm Fermium	Md Mendelevium	No Nobelium	Lr
				232.0381	231.03588	238.0289	(237)	(244)	(243)	(247)	(247)	(251)	(252)	(257)	(258)	(259)	(262)

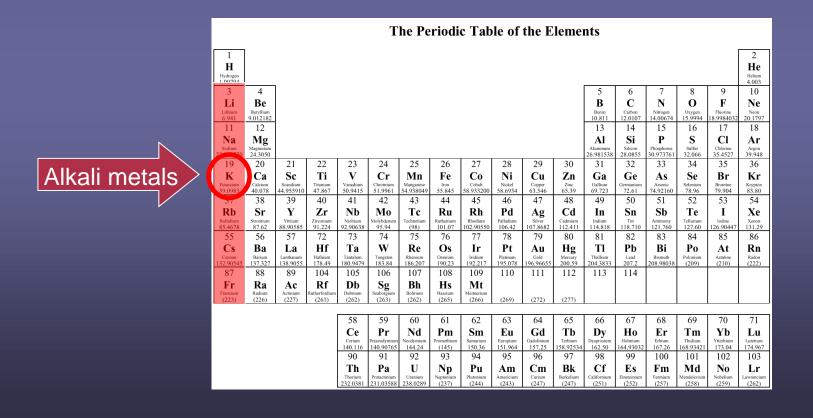
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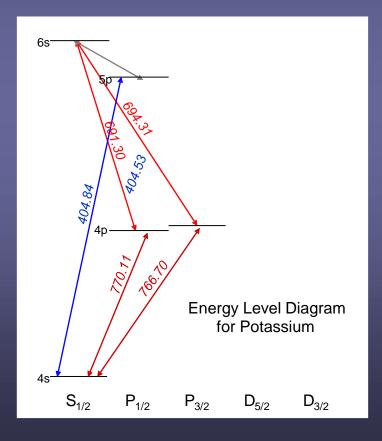
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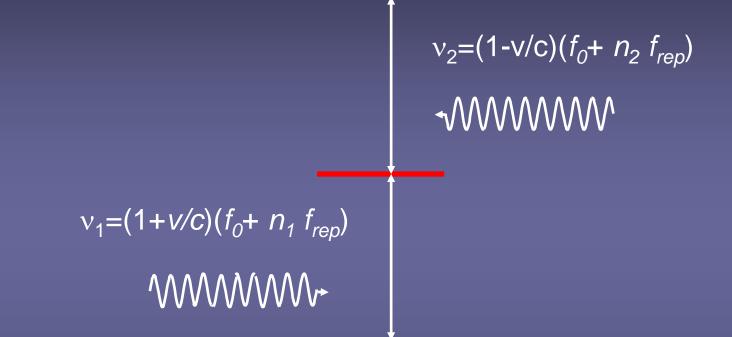
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What are we doing with potassium?

- Studying the spectral line shape of atomic potassium
- Collecting data and comparing to theory
- We can excite potassium to the 6s state via a "two-photon transition"

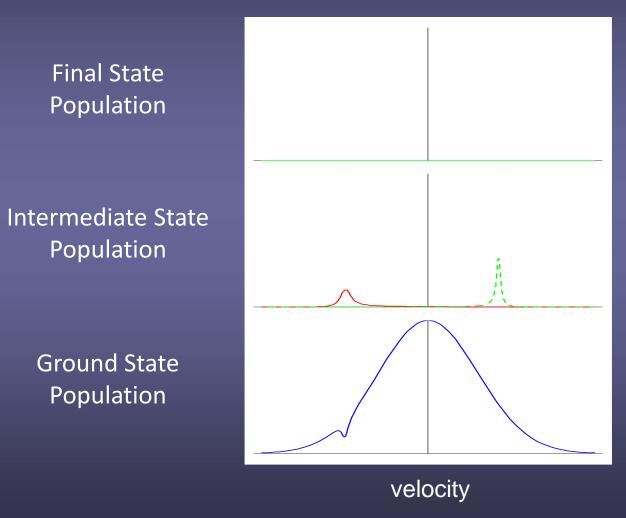


Velocity Selective Double Resonance



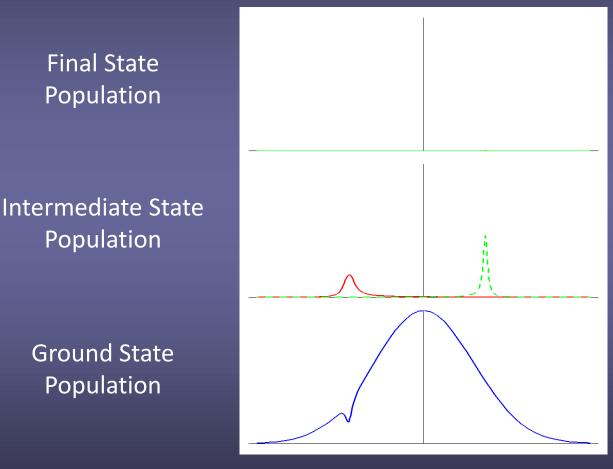
Both v₁ and v₂ depend f_{rep}.
Atoms are excited to the final state only when f_{rep} results in resonance for both transitions for a given speed.
Transitions can be excited from atoms with different speeds

Velocity selective double resonance



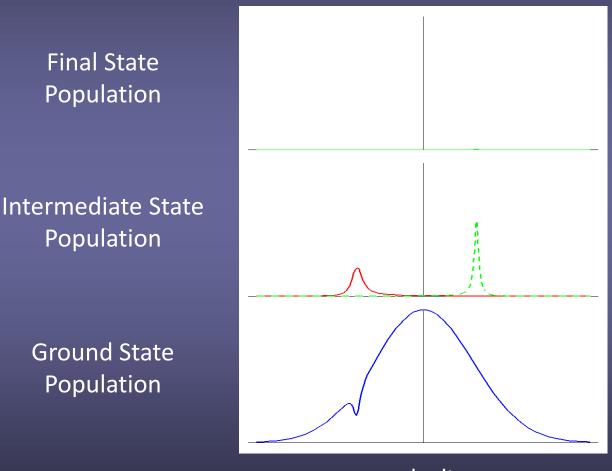
(Increasing f_{rep)}

Velocity selective double resonance

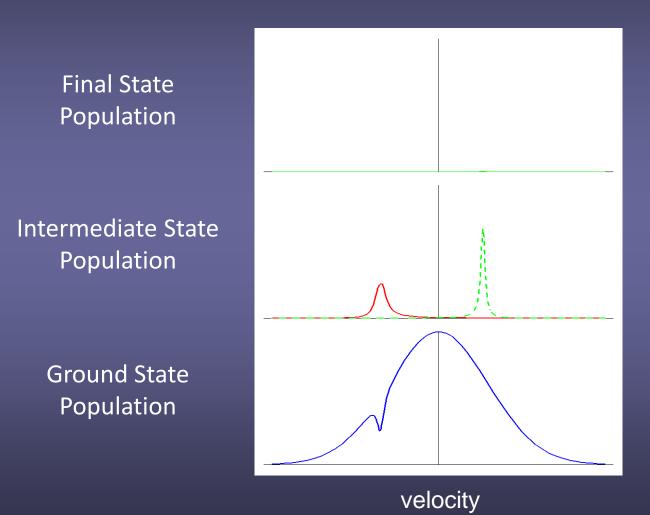


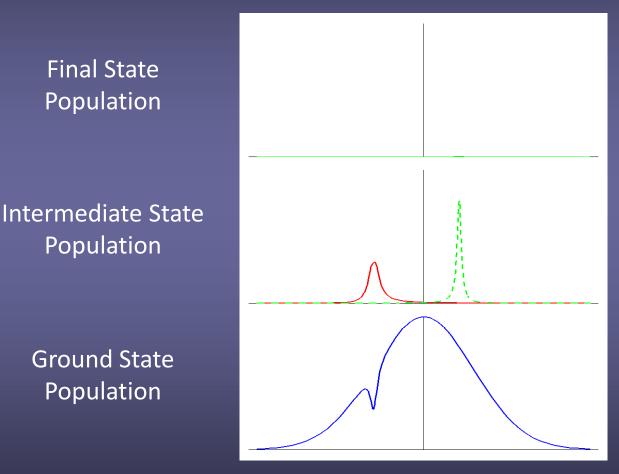
velocity

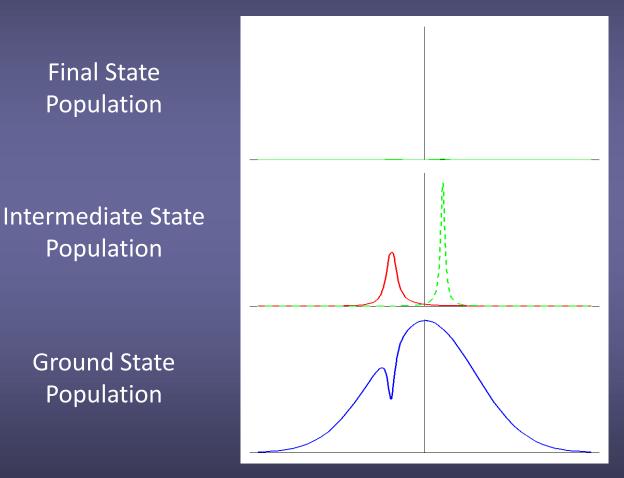
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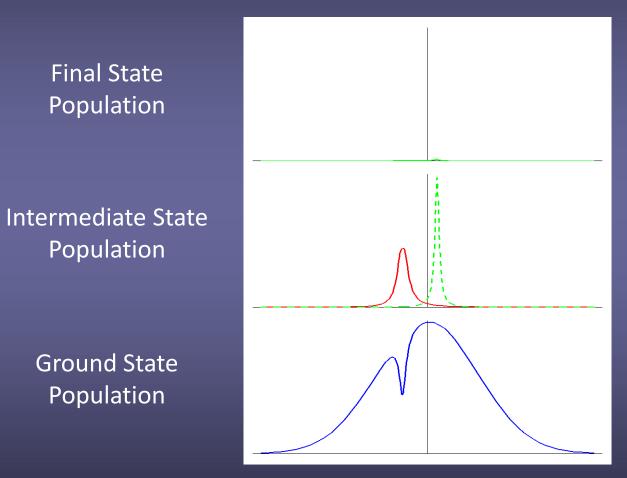


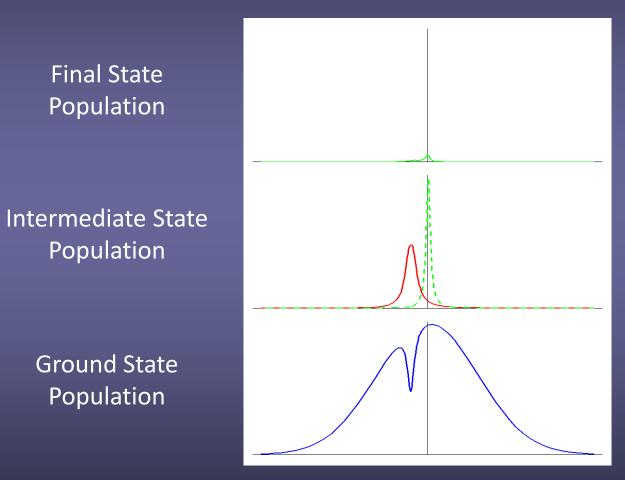
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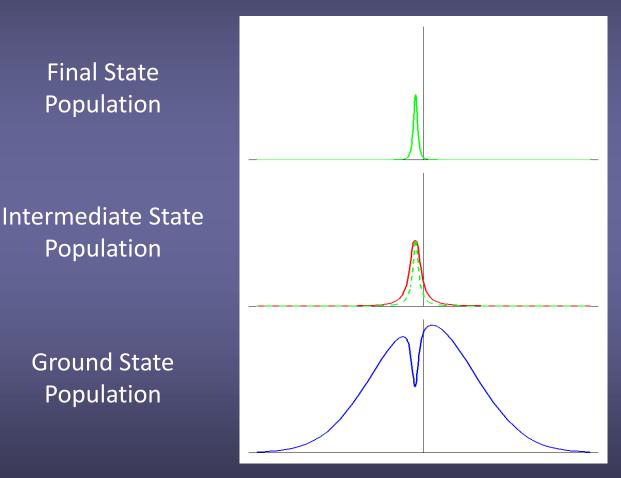


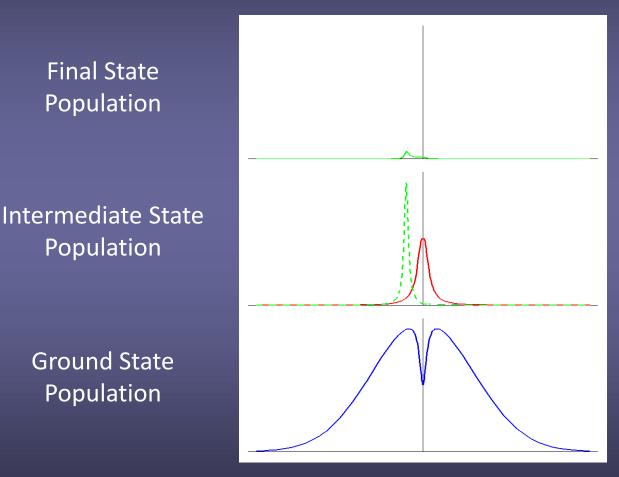


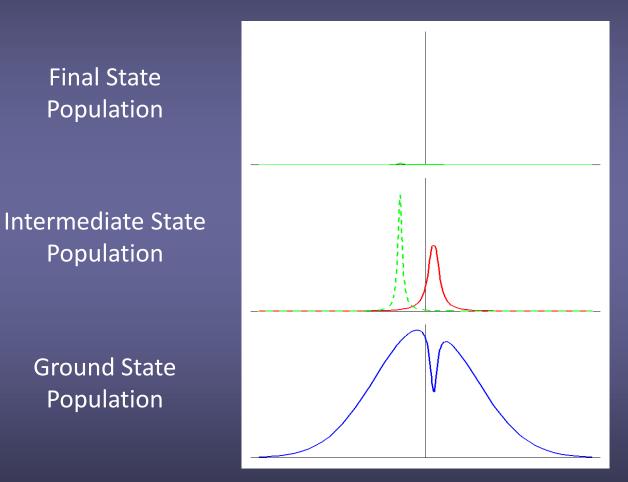


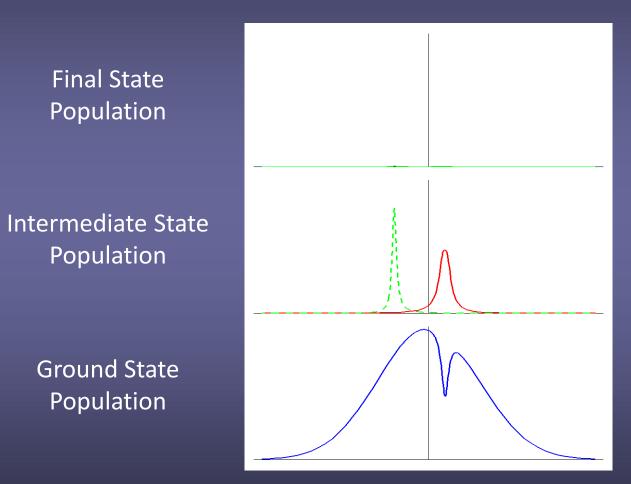


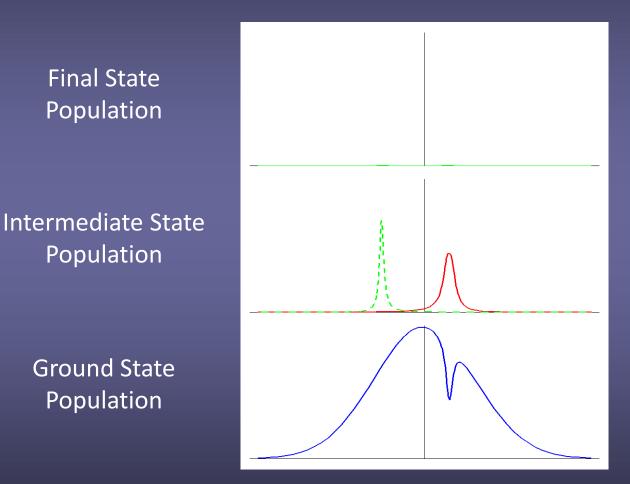


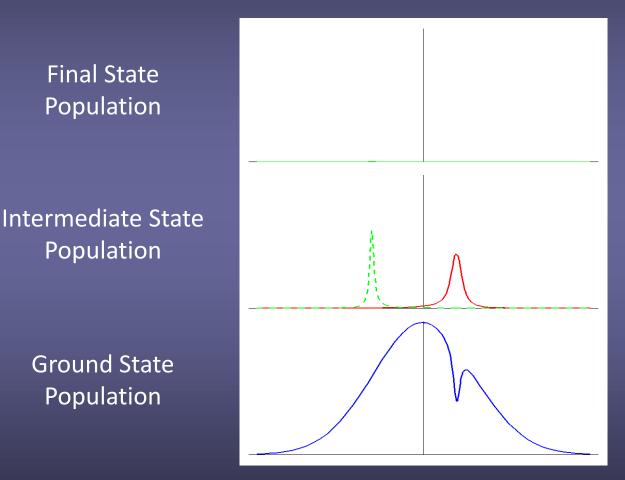


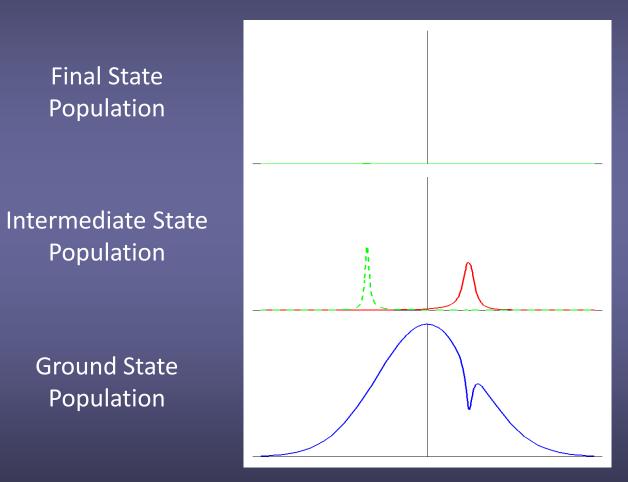


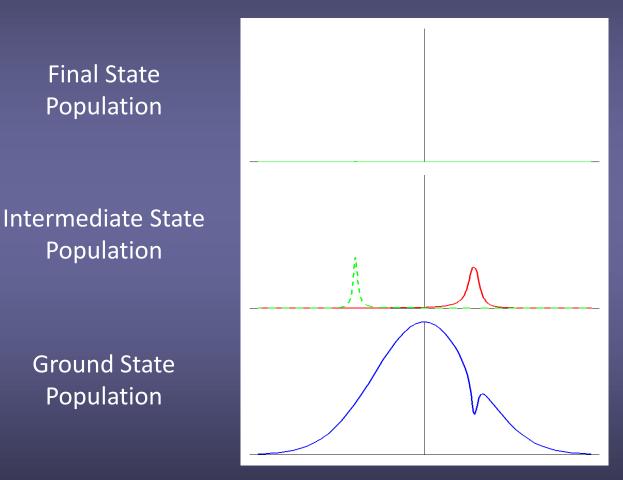


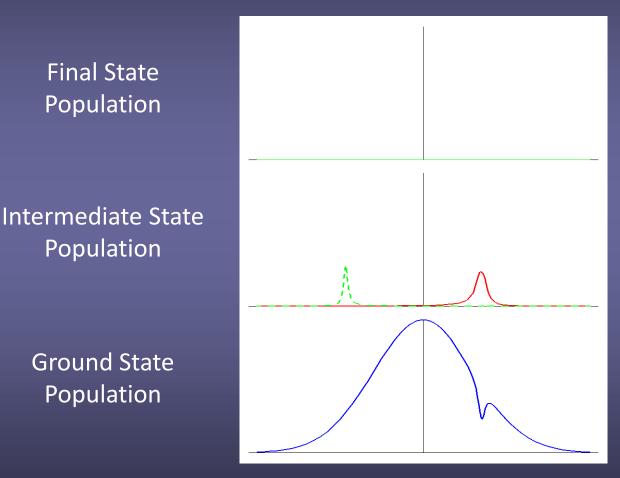


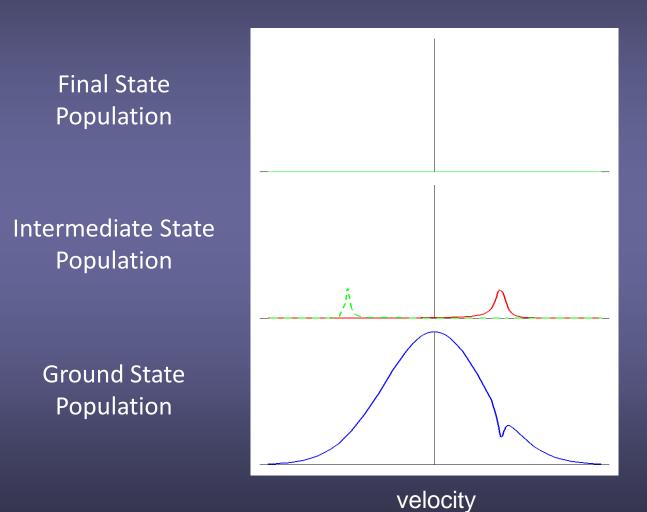


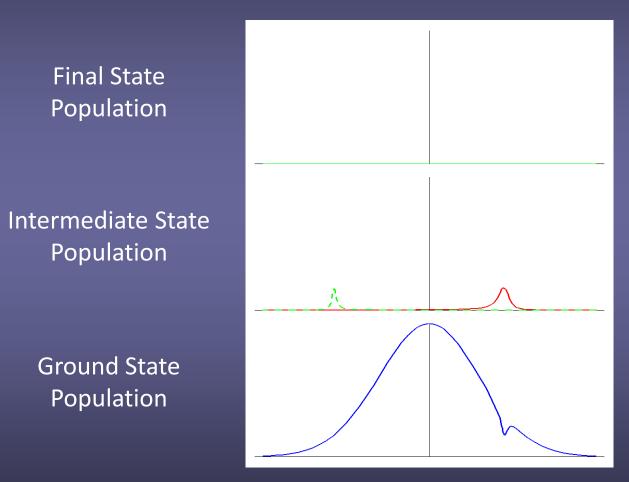


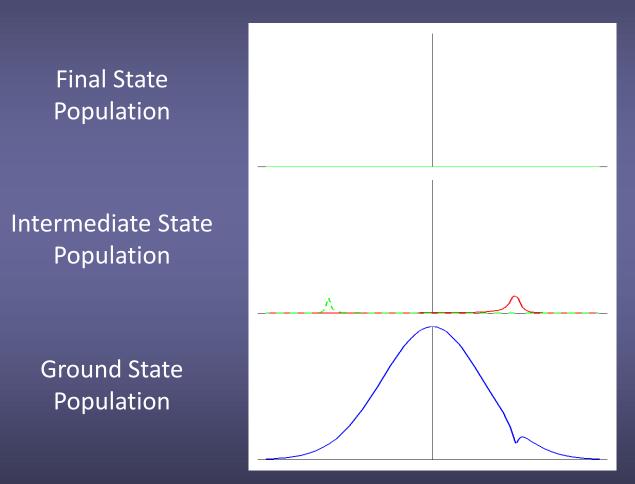




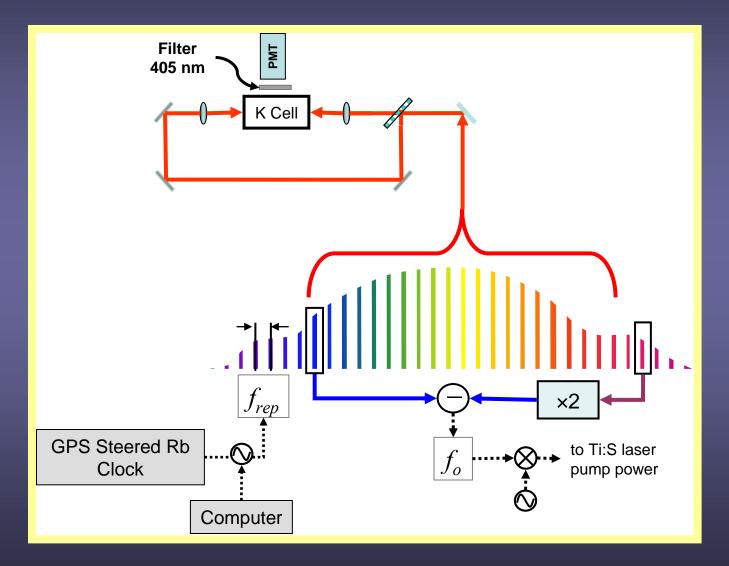




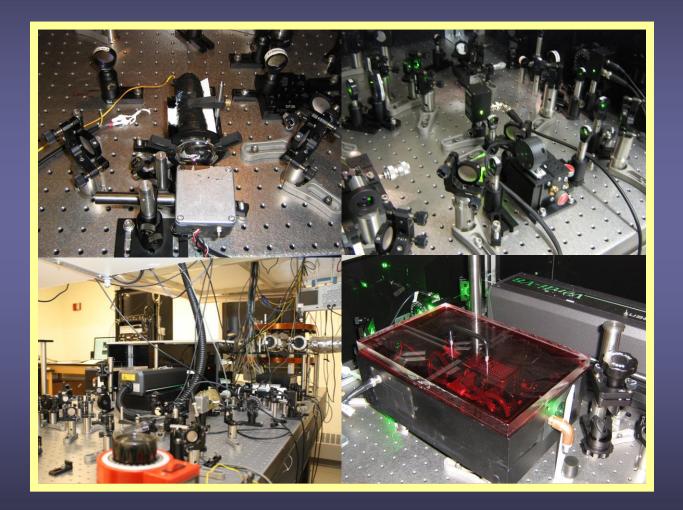




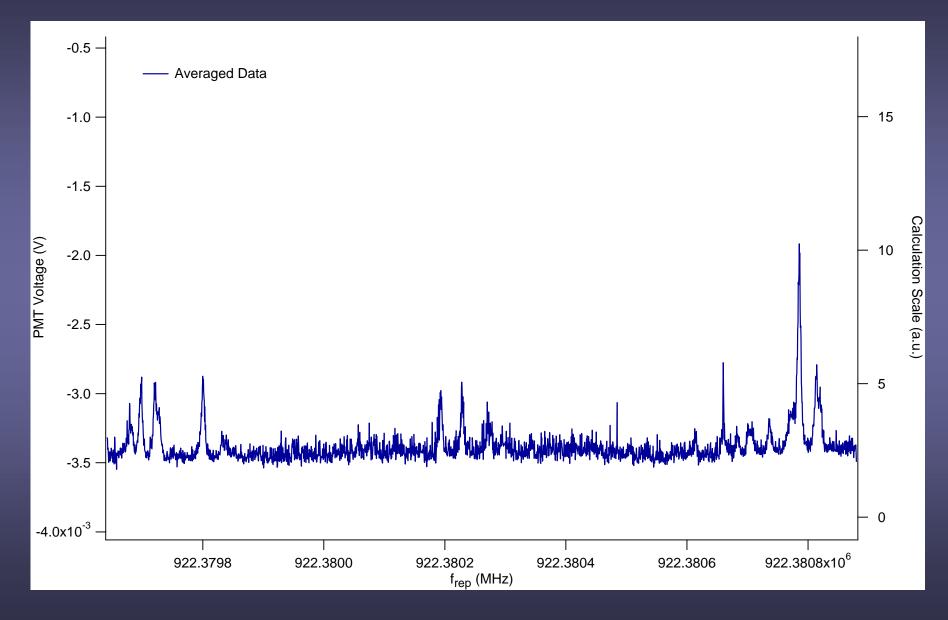
Experimental Setup



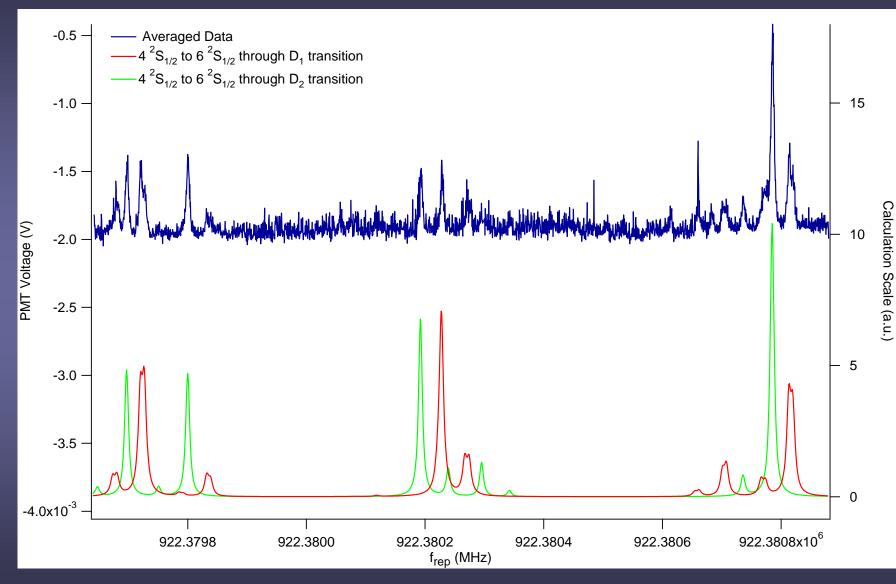
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Data!



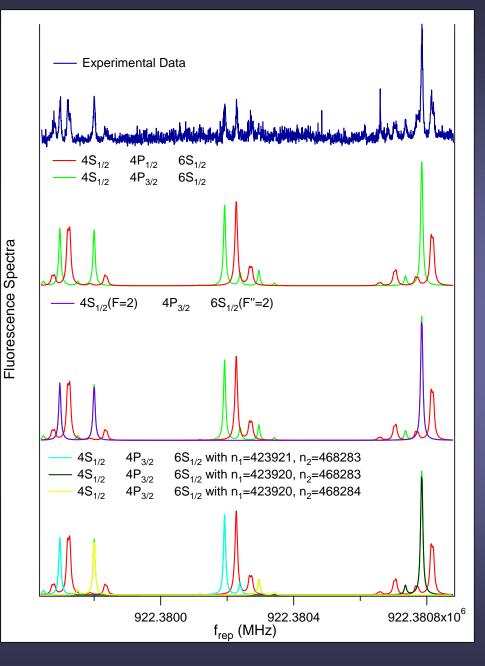
Data vs. Theoretical Modeling



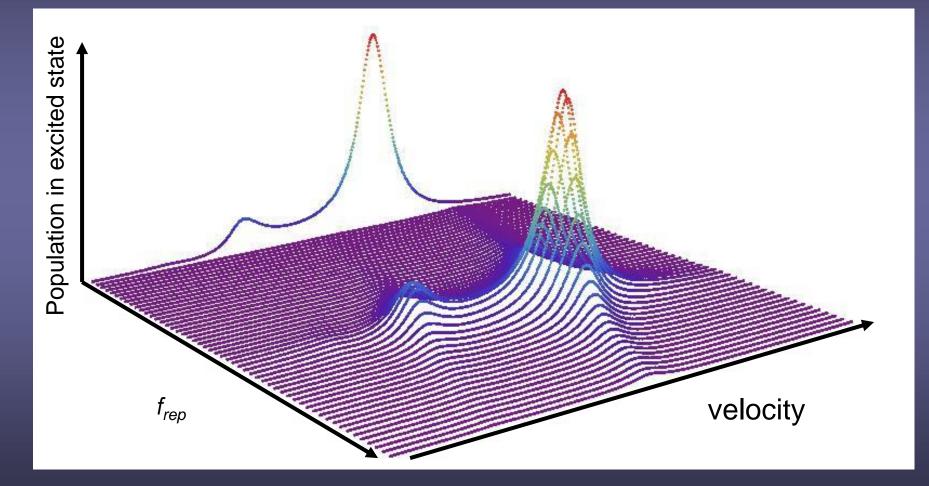
Though there are some rogue peaks, agreement is pretty good

More modeling

- Why do we have so many peaks?
 - In our models, we were able to "turn off" certain transitions
 - Allows us to isolate the peaks we observe in the data
 - Different velocity classes and mode pair combinations give rise to the same transition



Excitation of $4S1/2(F=2) \rightarrow 4P3/2 \rightarrow 6S1/2(F''=2)$ through Different Intermediate Hyperfine States



Acknowledgements

- Prof. Jason Stalnaker
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How do we make a frequency comb?

